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Preliminary communication

SYNTHESIS OF trans-AZOALKANE LIGANDS R-N=N-R BY DEHALOGENATION OF N,N-DIHALOALKYLAMINES, RNCl₂

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Summary

N,N-Dihaloalkylamines, RNCl₂, are dehalogenated by organometallic sodium or lithium salts to give *trans*-azoalkanes, R—N=N—R, and their corresponding metal complexes.

Dehalogenation of N,N-dihaloalkylamines, RNCl₂, to give azoalkane compounds, R—N=N—R, has not previously been observed. We now report that azocyclohexane and its complexes may be prepared in substantial yields from N,Ndichlorocyclohexylamine, C₆H₁₁NCl₂, by using sodium or lithium salts of organometallic complexes as dehalogenating agents.

When a THF solution of $Na_2[Cr_2(CO)_{10}]$ is allowed to react with $C_6H_{11}NCl_2$ at -40 °C the azocyclohexane complex $(C_6H_{11}N=NC_6H_{11})Cr(CO)_5$ (I) may be isolated in yields of 68%. I is also obtained from the reaction of $C_6H_{11}NCl_2$ with THF solutions of $(CO)_5Cr[C(OLi)(n-C_4H_9)]$ in yields up to 52%, whereas with the corresponding acylate complexes of Mo and W, $(CO)_5M[C(OLi)(n-C_4H_9)]$ (M = Mo, W), only poor yields of the azocyclohexane complexes $(C_6H_{11}N=NC_6H_{11})-M(CO)_5$ are obtained. Analogous syntheses of other azoalkane complexes starting from i- $C_3H_7NCl_2$ or n- $C_4H_9NCl_2$ have not yet been successful.

The yellow air-stable compound I melts at 96 °C with decomposition. In THF solution it dissociates completely to give $C_6H_{11}N=NC_6H_{11}$ and THFCr(CO)₅. This reaction is completely reversible and upon removal of THF I is recovered in quantitative yield. Similar but irreversible ligand exchange reactions occur with amines and phosphanes in benzene solutions at 20 °C to give the corresponding amine or phosphane adducts of $Cr(CO)_5$ in high yield.

The trans configuration of the azo ligand in I is proved by an independent synthesis of I from trans- $C_6H_{11}N=NC_6H_{11}$ and THFCr(CO)₅**. The trans structure

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^{**}This method was introduced by Dr. M. Herberhold for the synthesis of complexes of various *cis*azo ligands [1].

of the ligand in I has also been demonstrated by X-ray analysis.

The steric strain caused by the *trans* configuration of the bulky azo ligand may explain the reactivity of I, which makes it a useful reagent for the preparation of $LCr(CO)_5$ complexes.

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Reference

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